**AMMONIUM CHLORIDE**

### 1. Product Identification

**Synonyms:** Sal ammoniac; Ammonium muriate  
**CAS No.:** 12125-02-9  
**Molecular Weight:** 53.49  
**Chemical Formula:** NH₄Cl  
**Product Codes:** J.T. Baker: 0660, 0666,  
Mallinkrodt: 1614, 3355, 3363, 3364, 3384

### 2. Composition/Information on Ingredients

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>CAS No.</th>
<th>Percent</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ammonium Chloride</td>
<td>12125-02-9</td>
<td>99 - 100%</td>
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</tbody>
</table>

### 3. Hazards Identification

#### Emergency Overview

http://www.jtbaker.com/msds/englishml/A5724.htm  
11/21/2005

#### Potential Health Effects

**Inhalation:**  
Causes irritation to the respiratory tract. Symptoms may include coughing, shortness of breath.  
**Ingestion:**  
Causes irritation to the gastrointestinal tract. Symptoms may include nausea, vomiting and diarrhea.  
**Skin Contact:**  
Causes irritation to skin. Symptoms include redness, itching, and pain.  
**Eye Contact:**  
Causes irritation, redness, and pain.

### 4. First Aid Measures

- **Inhalation:** Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.  
- **Ingestion:** Induce vomiting immediately as directed by medical personnel. Never give anything by mouth to an unconscious person. Get medical attention.  
- **Skin Contact:** Immediately flush skin with plenty of water for at least 15 minutes. Remove contaminated clothing and shoes. Get medical attention. Wash clothing before reuse. Thoroughly clean shoes before reuse.  
- **Eye Contact:** Immediately flush eyes with plenty of water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get medical attention.

### 5. Fire Fighting Measures

- **Fire:**  
Not considered to be a fire hazard. At fire temperatures ammonium chloride begins to corrode metals and may dissociate into ammonia and hydrogen chloride. Mixtures of about 16% to 25% (by volume) ammonia gas in air are flammable.  
- **Explosion:**  
Not considered to be an explosion hazard.  
**Fire Extinguishing Media:**  
Use any means suitable for extinguishing surrounding fire. Water spray may be used to keep fire exposed containers cool.  
**Special Information:**  
In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode.

### 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Spills: Sweep up and containerize for reclamation or disposal. Vacuuming or wet sweeping may be used to avoid dust dispersal. US Regulations (CERCLA) require reporting spills and releases to soil, water and air in excess of reportable quantities. The toll free number for the US Coast Guard National Response Center is (800) 424-8802.

### 7. Handling and Storage

Keep in a tightly closed container. Protect from physical damage. Store in a cool, dry, ventilated area away from sources of heat, moisture and incompatibilities. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.

### 8. Exposure Controls/Personal Protection

#### Airborne Exposure Limits: Ammonium chloride:  
- ACGIH Threshold Limit Value (TLV):  
10 mg/m³ (TWA); 20 mg/m³ (STEL) Fume

#### Ventilation System:  
A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, *Industrial Ventilation: A Manual of Recommended Practices*, most recent edition, for details.

#### Personal Respirators (NIOSH Approved):  
If the exposure limit is exceeded and engineering controls are not feasible, a half facepiece particulate respirator (NIOSH type N95 or better filters) may be worn up to 50 times the exposure limit, or the maximum use concentration specified by the appropriate regulatory agency or respirator supplier, whichever is lowest. A full-face piece particulate respirator (NIOSH type N100 filters) may be worn up to 50 times the exposure limit, or the maximum use concentration specified by the appropriate regulatory agency, or respirator supplier, whichever is lowest. If oil particles (e.g. lubricants, cutting fluids, glycerine, etc.) are present, use a NIOSH type R or P filter. For emergencies or instances where the exposure levels are not known, use a full-facepiece positive-pressure, air-supplied respirator. WARNING: Air-purifying respirators do not protect workers in oxygen-deficient atmospheres.

#### Skin Protection:  
Wear impermeable protective clothing, including boots, gloves, lab coat, apron or coveralls, as appropriate, to prevent skin contact.

#### Eye Protection:  
Use chemical safety goggles and/or full face shield where dusting or splashing of solutions is possible. Maintain eye wash fountain and quick-drench facilities in work area.

### 9. Physical and Chemical Properties

#### Appearance:  
White powder.

#### Odor:  
Odorless.

#### Solubility:  
29.7g/100g water at 0°C (32°F)

#### Specific Gravity:  
1.53

#### pH:  
5.5 (1% aq.sol.); 5.1 (3% aq.sol.); 5.0 (10% aq.sol.)
10. Stability and Reactivity

Stability:
Stable under ordinary conditions of use and storage.

Hazardous Decomposition Products:
Involvement in a fire causes decomposition to form hydrogen chloride and ammonia.

Hazardous Polymerization:
Will not occur.

Incompatibilities:
Concentrated acids, strong bases, silver salts, potassium chlorate, ammonium nitrate, bromine trifluoride and iodine heptfluoride. Ammonium chloride reacts explosively with potassium chloride or bromine trifluoride, and violently with bromide pentafluoride, ammonium compounds, nitrates, and iodine heptfluoride. Explosive nitrogen trichloride may result from reaction of ammonium chloride and hydrogen cyanide.

Conditions to Avoid:
Heat, moisture, incompatibles.

11. Toxicological Information

Oral rat LD50 : 1650 mg/kg Investigated as a mutagen.

12. Ecological Information

Environmental Fate:
No information found.

Environmental Toxicity:
No information found.

13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

14. Transport Information

Not regulated.

15. Regulatory Information

16. Other Information

NFPA Ratings: Health: 2 Flammability: 0 Reactivity: 0

Label Hazard Warning:
WARNING! CAUSES IRRITATION TO SKIN, EYES AND RESPIRATORY TRACT. HARMFUL IF SWALLOWED OR INHALED.

Label Precautions:
Avoid contact with eyes, skin and clothing. Avoid breathing dust. Wash thoroughly after handling. Keep container closed. Use only with adequate ventilation.

Label First Aid:
In case of contact, immediately flush eyes or skin with plenty of water for at least 15 minutes. Remove contaminated clothing and shoes. Wash clothing before reuse. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. If swallowed, induce vomiting immediately as directed by medical personnel. Never give anything by mouth to an unconscious person. In all cases, get medical attention.

Product Use:
Laboratory Reagent.

Revision Information:
No Information Found.

Disclaimer:
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